

ITRCAL

Iteration of Calculated Nmr Spectra Using Least Squares Criteria
Instructions for Use

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ITRCAL

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I. INTRODUCTION

The analysis of complex coupled nmr spectra has been reported by Swalen and Reilley¹ and Castellano and Bothner-By,² as well as by Hoffman³ and Arata et. al.⁴ In all cases, a theoretical spectrum is calculated using standard techniques⁵, and then iterated for a best fit with a set of experimentally observed lines. In the ITRCAL program, the method of Castellano and Bothner-By is used. This program is the first reported implementation of the LAOCN3 algorithm on a minicomputer, and it should provide a powerful new scientific tool for laboratory real time analysis of experimental nmr spectra. ITRCAL requires a 12K Nicolet 1080 computer system and a NIC-294 Disk Memory.

The calculation of theoretical nmr spectra from chemical shifts and coupling constants has been well described in several places⁵⁻⁷ and will not be detailed here. Essentially, a high resolution Hamiltonian matrix is constructed as follows.

The diagonal elements are constructed according to the equation

$$H_{uu} = \sum_{i=1}^{n} \begin{bmatrix} v_i I_z(i) + \sum_{j=1}^{n} J_{ij} T_{ij} \\ j=1 \end{bmatrix}$$

where

vi is the chemical shift of nucleus i

 $I_z(i) = +1/2$ for spin α and -1/2 for spin β

 $\mathbf{J}_{ij}^{\mathbf{J}}$ is the coupling constant between nuclei i and j

 $T_{ij}^{1j} = +1/4$ if spins i and j of the basic function corresponding to that diagonal element are aligned and -1/4 if they are opposed.

The off-diagonal elements are calculated by

$$H_{uv} = 1/2J_{ij}U$$

where U = 1 if and only if basis function u and basis function v differ only in the interchange of spins i and j

and U = 0 otherwise

and J_{ij} is the coupling constant between the two spins which differ in the two basis functions.

Each such matrix is diagonalized to yield energy levels or eigen values as the diagonal elements. A matrix of eigenvectors is also produced by the diagonalization.

The energies of the transitions, in Hz, are calculated by subtracting each energy level in the previous H-matrix diagonal from each energy level in the

current H-matrix diagonal. The probability or the intensity of these transitions is given by

$$I_{km} = \begin{pmatrix} n & n \\ \Sigma & \Sigma & C_{ik}C'_{jm}X_{ij} \end{pmatrix}^{2}$$

$$i=1 \quad j=1$$

where

I_{km} is the intensity of the transition between energy level k of the last H-matrix and energy level m of this H-matrix

 C_{ik} is the i^{th} coefficient of the k^{th} column of the last set of eigenvectors

 C'_{jm} is the jth coefficient of the mth column of this set of eigenvectors $X_{ij} = 1$ if and only if basis function i of the last set and basis function j of the current set differ only in one spin

= 0 in all other cases.

Iteration for a best fit between a theoretical and an experimental spectrum requires that a set of close guessed parameters be obtained from which to iterate. The actual iteration procedure is a least squares method which tends to minimize the rms error. The amounts which parameters are varied is calculated from a set of partial differentials which are used to minimize the differences between the observed and calculated frequencies. This is discussed in detail in the paper by Castellano and Bothner-By.²

Loading and Startup

To load ITRCAL, place the tape in either the high speed or low speed reader, start the DEMON monitor and call the binary loader by typing BIN. The loader will be loaded and started automatically, and if a high speed reader is in use, the tape will begin reading in. If a low speed reader is in use, turn it on after calling BIN and the tape will begin to read in.

When the tape has read in to the rubout in the trailer section of the tape, the computer will stop. Then, restart the monitor at 7600 and store the memory blocks as follows:

STORE ITRCL1 0-7577: P

STORE ITRCL2 114225 - 115777: P

Then, if you wish to make use of the 7-spin calculation feature, load the NMRCL7 overlay tape and read it in by typing BIN again. When this tape has read in, the computer will stop. Restart DEMON at 7600, and store this program by the following commands:

STORE NMRCL1 0-7577: P

STORE NMRCL2 115461 - 115656: P

To start ITRCAL, load the two ITRCAL files by the following commands:

LOAD ITRCL1

LOAD ITRCL2

GO

The GO command will start the program at location 0, and cause it to type out its title and begin accepting data. Note that the RUN command of DEMON II cannot be used, since non-contiguous storage is involved. Once ITRCAL has been used, it modifies its start-up routine so that it no longer goes through the questions for the calculation of a spectrum before awaiting additional commands. The program can then be restarted using the R command.

Initial Dialogue

Upon startup the first time or whenever the program is restarted, ITRCAL will ask the following questions to obtain parameters for the calculation of a spectrum.

ITRCAL V1B 04/73

NUMBER OF SPINS =
SPECTROMETER FREQ (MHZ) =
WIDTH OF DISPLAY (HZ) =
OFFSET OF DISPLAY (HZ) =
INTENSITY THRESHOLD =

~ 0.01

(Aufang

This is the beginning of an actual calculation of a guessed or theoretical spectrum. ITRCAL must have an approximate spectrum from which to iterate for a best fit with the observed spectrum. The display frequency convention is that of all other Nicolet programs: the right end is considered to be 0 Hz and the frequency increases in the positive direction to the left. This conforms to the most common scale for both pmr and cmr measurements.

The intensity threshold is the minimum intensity transition that is to be saved in the table of transitions and intensities. Up to 300 transitions can be saved; however for cases involving fewer transitions this threshold can be entered as zero. If no number is entered and a Return is typed, the last entered value for the threshold is assumed. For cases involving 2, 3 or 4 spins, no threshold is necessary.

The spectrometer frequency is used if and only if chemical shifts are entered in ppm or tau units, and is otherwise ignored completely. In the case where shifts are entered only in Hz, the spectrometer frequency may be set to zero by simply typing Return.

The entire spectrum is calculated regardless of the setting of the display region, and the offset and width may be changed after the calculation to examine any portion of the spectrum desired. If no values are entered for offset and width, an offset of zero and a width of 2048 are assumed.

Entry of Shifts and Coupling Constants

The program will next ask for the number of chemical shifts and coupling constants which would be expected from the number of spins which were entered. For example, if $\underline{3}$ is entered for NUMBER OF SPINS =, the program will type:

CHEMICAL SHIFTS

V(1)=

V(2) =

V(3)=

COUPLING CONSTANTS

J(1,2) =

J(1,3)=

J(2,3) =

The chemical shifts are assumed to be in Hz if their value is terminated with a Return. If only a Return is entered, a zero value is assumed. If the chemical shift is terminated with a \underline{P} , the shift is assumed to be in parts per million, where the right end is zero Hz and the ppm scale increases positively downfield. If a chemical shift is terminated with a \underline{T} , the tau scale is assumed, in which TMS is assumed to be at 10τ , and the spectrum decreases in positive value to the left. If no spectrometer frequency has been entered, the use of the tau or ppm scales will result in all shifts being assigned to zero Hz. The coupling constants, of course, are always entered in Hz.

Following the entry of all shifts and coupling constants, the program will type:

SET HORIZ DISPLAY SCALE TO 4K
SET VERT DISPLAY SCALE TO 1K (READJUST AS NEEDED).

On 1080 data systems these operations should be performed before proceeding. On BNC-12 and NMR-80 systems, simply set the Size buttons to 4K and the vertical display scale to 1K. The program will print a > sign and is then ready to begin calculation of the spectrum. Any of the commands described below can then be entered.

II. DESCRIPTION OF BASIC COMMANDS

Each command is one character in length and is executed when a Return is typed. The program is ready for a new command when a new > is printed. The commands may be entered in a string and are executed by typing one return after the entire string. Each command letter is printed and the command is then executed. When all commands are completed a new > is printed. If an illegal command is typed, such as X, the program will print

X IS ILLEGAL

and, if a command string has been entered, go on to the next command. For example, if the string VVVXJ is entered, followed by a Return, the program will allow entry of three V chemical shifts, type X IS ILLEGAL, and allow entry of one coupling constant. A number of commands can be stopped in midstream by typing \underline{Q} . This will always produce the question

ABORT QUEUE?

If \underline{Y} is answered, the rest of the current command string is also aborted. If \underline{N} is answered, the rest of the string is completed. If no string exists at the time the answer is immaterial.

ITRCAL Commands

The following commands are legal in ITRCAL:

- A Assign previously unassigned transitions
- B Begin iteration sequence again
- C Change a transition assignment
- E Print energy levels during next calculation
- G Go perform a non-iterated calculation
- I Iterate
- J Change a J-value
- K Print out the current shifts and coupling constants
- L Lorentzian line shape calculation
- Monitor call jumps to disk monitor
- N Enter Nicobug II
- O Change Offset
- Q Quit current routine and return to command decoder
- R Restart program at beginning clear all arrays
- T List Transitions on Teletype
- V Change chemical shift
- W Change sweep width
- Z Zero all transition assignments

The commands D, F, H, P, S, U, X and Y are illegal

G - Go Calculate a Spectrum

The G command takes the currently entered values of V and J, and calculates the stick spectrum. The display is active, showing the lines produced at the end of each calculated matrix, so that one can actually watch the spectrum grow as it is calculated. As calculation proceeds, the numbers of the individual matrices are printed, giving the user a feeling for how far along the calculation has proceeded. For N spins, there are always N+1 matrices, so for example a 3 spin case would print out

G 0 1 2 3

If a degeneracy is encountered, the degenerate energy levels are printed out during the calculation.

L - Calculate a Lorentzian line shape for the stick spectrum

After L is typed, the program will print

LINE WIDTH =

and allow entry of the line width of Hz. If a 0 is entered, the stick spectrum will be displayed. If any positive non-zero number is selected, the program will begin calculating the line shape, displaying the result as each line in the spectrum is calculated.

J - Change a coupling constant

The J command causes the printing of J followed by a left parenthesis. The program awaits entry of a number designating the actual nuclei between which coupling occurs. The nucleus numbers must be entered in ascending order. The program will not permit illegal numbers to print when they are typed. For example, to change $J_{2,3}$ you would type only the J, the 2 and the 3. The punctuation is added by the program producing J(2,3)=. Any number can then be entered.

V - Change a chemical shift

The V command causes the printing of a $\underline{V}(.)$ The user enters the number of the nucleus whose shift he wishes to change, followed by the actual value of the shift. Thus, in entering V(3)=118.2, the user types V, 3 and 118.3 and the program types $\underline{(, and)}=.$ As before, shifts can be entered in tau or ppm units by terminating the entry with a T or P.

O - Change display offset, W - change the display width

The offset and width of the display can be changed using these commands; if an entire recalculation is performed following this change, the complete spectrum of any region may be displayed in this way. However, the offset and width change the display immediately after the entry of these commands, and the new display will be

constructed from the stored transitions. Since only 300 transitions are stored, the display may not appear exactly correct in the case of 5, 6 or 7 spin systems. The offset and width cannot be changed unless ITRCAL has performed a calculation (G or I) since it was loaded.

E - Print energy levels

This command turns a "switch" on and off controlling the printing of energy levels during a calculation. When E is typed the first time it turns this switch on, and when it is typed again, it turns it off. When the E switch is on, the energy levels are printed out as calculated as the eigenvalues of each set of matrices. These values are not saved between matrices so they are not sorted by increasing value.

R - Restart program at beginning

This command causes a complete restart of the program. All shift and transition arrays are cleared. It is entirely equivalent to pushing Stop and then starting at address 0.

M - Monitor call

M calls the disk monitor by jumping to 7600.

N - Call Nicobug II

 \underline{N} causes a jump to Nicobug II, at 114700. This program is contained in the ITRCAL tape and is always present for your use. It should only be required for actual user modifications of ITRCAL.

T - List Transitions on Teletype

This command will cause the actual listing of all of the saved transitions and their intensities. Transitions above the selected threshold will have been saved. Only 300 transitions can be stored or listed. The listing can be stopped by typing Q.

Q - Quit current routine

The \underline{Q} command will stop the transition listing and will also allow exit from the iteration and transition assignment modes.

III. COMMANDS USED IN ITERATION

I - Iterate

The Iteration command assumes that a calculation is to be made using guessed, chemical shifts and coupling constants, followed by an assignment of actual observed frequencies for least squares iteration. Following program startup, the V and J values are entered. Instead of actually calculating the spectrum by typing G, typing I will allow spectrum calculation as well as iteration.

Immediately after the I command is entered, the questions

DESIRED VALUE OF RMS ERROR = MAXIMUM # OF ITERATIONS =

are printed. Answer the rms error question with some stall number, such as 0.1. This number can be varied empirically after some experience is gained with a particular case. If no value is entered, the last entered value is assumed. The ITRCAL iteration technique converges very strongly if it converges at all, and probably no more than 5 to 10 iterations will ever be needed.

When the above two questions are answered, the messages

ENTER VARIABLES

SET 1:

SET 2:

etc.

are printed. These variable sets are those groups of V's and J's which are to be varied. If no sets are entered, no iteration is performed since no parameters are to be changed. In the case of symmetrical molecules, several shifts or coupling constants might be varied as a group so that they will not become non-equivalent by iteration. In this case, all those which are to be varied together are entered on the same line. It is not possible to enter both V's and J's as part of the same set. They must be entered as separate groupings.

For example, in iterating an A_2X system, the parameters to be varied are V(1), V(2), V(3), J(1,2), J(1,3) and J(2,3). If all of these are to be varied separately, they are entered on separate lines as different variable sets. If, however, nuclei 1 and 2 are actually both A nuclei, they should be varied together. Furthermore, J(1,3) and J(2,3) would then also be varied together. It should also be noted that in this case, J(1,2) would probably be zero and should not be varied at all.

To accomplish this, the following set assignments are made:

ENTER VARIABLES

SET 1: V(1) V(2)

SET 2: J(1,2) J(2,3)

SET 3: V(3)

SET 4:

Terminate each entry with a Return. To terminate the list of sets, type Return without entering any value for the last set. The program will ask

ARE YOU SURE?

This gives you a chance to inspect the list for omissions, redundancies and incorrect groupings. If none are found, type \underline{Y} followed by a Return. If some are found, typing any other character will allow you to enter a new list. Once you are sure, the program checks for redundancies, such as having the same parameter as a member of two groups. Should it find such an error, it will type a question mark, ring the Teletype bell and print

**INCONSISTENCY!

This message has been generally found to be emphatic enough to catch the user's notice.

Once a consistent set of variables has been entered, the iteration process will begin. Iteration 0 is simply a calculation of the spectrum based on the guessed J's and V's. The program prints

ITERATION 0: 0 1 2 3

for the matrices involved in the calculation. It then begins the process of transition assignment. This is done by printing out the calculated transition and intensity and allowing entry of the observed value. The observed values are those actually measured in your nmr spectrum either by peak printout or by ruler measurement. A typical line assignment is shown below:

7 8 8.671 0.656 =

The numbers are the energy level numbers between which the transition occurs, the calculated frequency and its intensity. The observed value is entered following the equals sign. It is not necessary to assign all the lines in a spectrum. In fact, it is generally more desirable to assign the strong well-defined lines and omit the less intense ones. It is desirable to use intensities as a guide in line assignment, since the actual frequencies may be out of order because of degeneracies. Since the least squares analysis is usually performed on an overdetermined system, only some lines need be assigned.

The assignment questions can thus be answered not only with a line frequency but also with a Return, meaning do not assign this line, or with a "mark, meaning assign this line the same frequency as the line above.

When all frequencies have been assigned, the iteration will begin. For each iteration, the program will print out the iteration number and display the number of lines calculated after each matrix is complete. Iteration will halt when (a) the rms error is less than the desired value, (b) when the change in rms error between iteration is less than 3%, or (c) when the number of iterations is exhausted. Iteration can be halted at any time using the Q command. When iteration stops, the program will print out the rms error and the calculated chemical shifts and coupling constants.

T - Type out the transitions and intensities

In the iterated case, when the $\underline{\mathbf{T}}$ command is given, the final iterated transitions will be printed out in ascending order, with their assigned frequencies printed in the last column for examination.

A - Assign previously unassigned transitions

If rough convergence is not sufficient, it can sometimes be improved by assigning the weaker transitions and continuing with the iteration. This command will print out the value of all unassigned transitions and allow assignments. The \underline{I} command will then allow additional iteration.

K - Print out current chemical shifts and coupling constants

This command simply lists the current V's and J's for inspection. It indicates what values the program will iterate from and how far from the original guessed values they have been changed.

C - Change a transition assignment

The \underline{C} command allows changing of any given transition assignment. This can be done by typing \underline{C} followed by a Return. The program will await entry of the energy level numbers of the transition whose assignment is to be changed. Each level should be followed by a space. When two level numbers have been entered, the program types an equals sign and awaits entry of the new value. Typing \underline{Q} will allow exit without entering any value if the wrong transition was inadvertently chosen.

B - Begin iteration sequence again

Preceding the \underline{I} command with a \underline{B} command allows the program to start at the beginning of the iteration sequence. It zeroes the iteration counter, allows entry of a new rms error and iteration count and new parameter sets.

Z - Zero all transition assignments

This command allows the zeroing of all transition assignments. Thus, ZA strung together would allow a complete new entry of all observed frequencies.

Plotting

Because of the extreme core limitations imposed by writing the ITRCAL program to perform in a 12K minicomputer, no plotting routine is provided. However, both the stick and Lorentzian spectra can be plotted using routines in the FT-Nmr and T1PRGM packages. The Lorentzian spectrum is stored in the first 1K of data memory and the stick spectrum in the seventh 1K of data memory. Plotting of the stick spectrum can best be accomplished using the interpolating plot routines. It will produce extremely narrow lines.

In going to and from the plot routines, it is sometimes desirable to have ITRCAL remember its current values. To do this, store ITRCAL upon re-entry to the monitor as follows:

>M
*STORE ITRCL1 0-7577: P
REPLACE? Y

(Set Display to Starting = 0 and Size = 8K)

*STORE ITRCL2: B
REPLACE? Y
*LOAD T1PRGM
*GO
& PL

Error Modes

If ITRCAL finds itself performing an "impossible" calculation, such as division by zero, infinite looping, or inversion of a singular matrix, it may exit from the program to Nicobug II, which is stored at 114700. It will print out

ERR @ nnnn

where nnnn indicates the address of the calculation inconsistency. ITRCAL will not "self-destruct," however and can be restarted from Nicobug by typing \underline{G} .

IV. EXAMPLES OF USES OF ITRCAL

A. An ABX System

Let us consider a simple example of a spectrum for iteration. We will analyze the simple ABX system shown in Figure 1. This spectrum produces the FT-Nmr peak printout which is shown below, and we can, from this information, deduce sufficient information for some guessed parameters to enter in ITRCAL. We can also use this frequency information to enter as the observed transition values during iteration.

• PP				
136	78.1425	0.8682	PPM	284
214	71.2864	0.7920	PPM	480
228	70.0558	0.7783	PPM	654
306	63.1996	0.7022	PPM	828
553	41.4885	0.4609	PPM	444
611	36.3903	0.4043	PPM	1000
645	33.4017	0.3711	PPM	236
704	28.2157	0.3135	PPM	468
793	20.3926	0.2265	PPM	846
851	15.2944	0.1699	PPM	420
870	13.6244	0.1513	PPM	596
928	8.5262	0.0947	PPM	266

It is always necessary to have some sort of external information about the compound in order to perform such an iteration. For instance, coupling constants and chemical shifts of simpler, model compounds are always of a great deal of value in deconvolving a closely coupled spectrum. In this case, however, we will assume that we have little or no such information other than that shown in the peak printout.

Actually, of course, one can analyze an ABX system by a series of complex mathematical manipulations, such as are described by Pople⁵ and Becker⁶. However, with ITRCAL, we will discover that most of this complexity becomes unnecessary, as the simplest guesses will be sufficient for analysis.

We will first extract the information that comes easily from ABX spectra. The values ν_X and J_{AB} are easily obtained. ν_X is simply the center of the X multiplet, and can be found by adding half the line separation to the lower frequency lines. Thus,

$$v_{\rm X}$$
 = (78.1425 - 63.1996)/2 + 63.1996 = 70.67
or
 $v_{\rm X}$ = (71.2864 - 70.0558)/2 + 70.0558 = 70.67

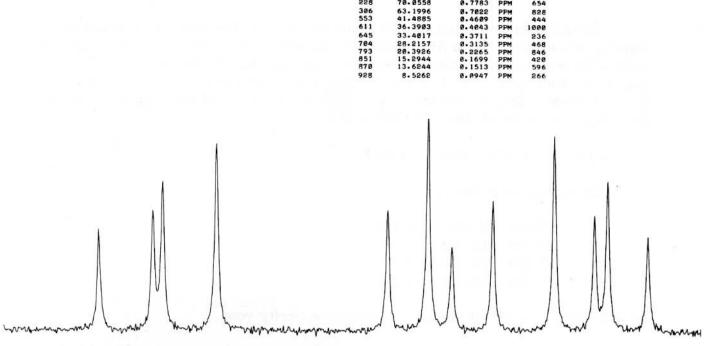
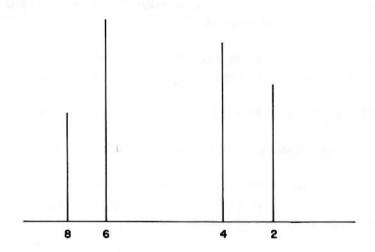


Figure 1. Experimental ABX spectrum



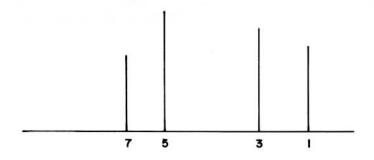


Figure 2. The two quartets in the AB part of the ABX

The AB coupling constant JAB is almost as easy to obtain. It is found by dividing the AB part of the ABX spectrum into two symmetrical quartets. This can usually be done by inspection. The ones of one quartet are labelled 1, 3, 5 and 7 and of the other quartet 2, 4, 6 and 8. Note that these lines may or may not be exactly alternate. This is illustrated in Figure 2. It turns out that J_{AB} is given by the differences between lines in the quartets, or

$$J_{AB} = 3 - 1 \text{ or } 4 - 2 \text{ or } 7 - 5 \text{ or } 8 - 6.$$

Thus, JAB is given by

13.624 - 8.526 = 5.098 or 20.392 - 15.294 = 5.098 or 33.401 - 28.215 = 5.186 or 41.488 - 36.390 = 5.098

We will adopt $J_{AB} = 5.098$ by overwhelming majority vote.

Now the remaining parameters, J_{AX} , J_{BX} , v_A and v_B are considerably harder to extract. Instead of plowing through the tedious equations necessary to find them, we will simply make some semi-naive, semi-educated guesses and try to iterate towards a spectrum of similar character.

As a first guess, let us simply take the center of gravity of the two groups of four lines (not of the two quartets mentioned earlier) as v_A and v_B .

We will simply call
$$v_A$$
 = (15.294 - 13.624)/2 + 13.624 = 14.459 v_B = (36.390 - 33.401)/2 + 33.401 = 34.896

These will at least allow calculation of some spectra to see if we can come close. As our second naive assumption, we will simply set J_{AX} and J_{BX} to 5 Hz, too. Then, let us calculate the spectrum and see what we find.

The actual parameters entered are shown below.

```
*LOAD ITECL1
*LOAD TIRCL?
*60
                94/73
I TRCAL
         VIB
NUMBER OF SPINS= 3
SPECTROMETER FREG (MHZ)=
WIDTH OF DISPLAY (HZ)= 90
OFFSET OF DISPLAY (HZ)=
INTENSITY THRESHOLD= 0
CHEMICAL SHIFTS
V(1) = 14.459
V(2)= 34.996
V(3) = 70.67
COUPLING CONSTANTS
J(1,2)= 5.098
J(1,3) = 5
J(2,3) = 5
SFT HOEIZ DISPLAY SCALE TO 4K.
SET VERT DISPLAY SCALE TO IN (READJUST AS NEEDED).
>GL
G 0
     1
         2
            3
LINE WIDTH= .5
*LOAD TIPEGM
*G0
```

The actual calculated spectrum is shown in Figure 3 and obviously bears little resemblance to our experimental one. It is, in fact, one of the "deceptively simple" ABX spectra which the literature always warns us about.

& PL.

For our next least naive assumption, we will simply assume that we can approximate J_{AX} and J_{BX} from the spacing of the quartet. Thus, we will call both J_{AX} and J_{BX} about 7 Hz. There is no need to supply further significant figures, since these are just guesses. The parameters entered below produce the spectrum shown in Figure 4.



Figure 3. Deceptively simple ABX produced by injudicious choice of parameters.

```
ITRCAL VIB
               74/73
NUMBER OF SPINS= 3
SPECTROMETER FREC (MHZ)=
WIDTH OF DISPLAY (HZ)= 90
OFFSET OF DISPLAY (HZ)=
INTENSITY THRESHOLD = 0
CHFMICAL SHIFTS
V(1) = 14.459
V(2)= 34.896
V(3) = 70.67
COUPLING CONSTANTS
J(1,2)= 5.09R
J(1,3) = 7
J(2,3) = 7
SET HORIZ DISPLAY SCALE TO 4K.
SET VERT DISPLAY SCALE TO 1K (READJUST AS NEEDED).
>GL
  0
      1
         2
            3
LINE WIDTH= .5
> M
M
*LOAD TIPRGM
*60
& PI.
```

This last spectrum looks pretty close to the experimental one. In fact, the line shift differences are only minimal and their ordering about the same. We need only go into iteration mode to calculate the best possible spectrum fit. We should emphasize again here, that the assignment of lines is not necessarily a one-to-one procedure. The guessed spectrum may look similar but have some of the lines out of order. In this case, it is important to go by relative intensities rather than by strict frequency order.

Let us now proceed to iteration. The actual program output is shown on the facing page. The guessed values for the chemical shifts and coupling constants are entered and the Iterate command is given. Note that one can start at the iterate command if a theoretical spectrum has been calculated with the parameters which are to be the guessed ones for iteration. An rms error of .01 is selected along with a maximum iteration count of 10. The variables are entered as six separate groups and confirmed by answering Y to ARE YOU SURE?.

The first calculation proceeds and lines are assigned to all theoretical transitions. Iteration continues with an rms error which decreases during the first three iterations and which stabilizes at .0254 after three iterations. Since there is no change after this many iterations, the program stops and lists the chemical shifts and coupling constants that have been calculated. A line width of 0.5 Hz is added and the calculated vs. observed transitions are printed out. Finally, the Monitor is called and the resulting spectrum plotted out using the PL routine of T1PRGM. The calculated spectrum is shown in Figure 5. The iteration has clearly converged to a meaningful solution, since the spectra are virtually superimposable. The actual spectrum used was generated by adding noise to a theoretical spectrum having the values

V(1) = 15 V(2) = 35 V(3) = 70 J(1,2) = 5 J(1,3) = 7J(2,3) = 8

so the calculated values clearly agree quite well.

It should be noted that the signs of coupling constants in an ABX are not determined and that the same spectrum could be produced using some negative coupling constants. The actual values used must be selected with careful attention to chemical criteria.

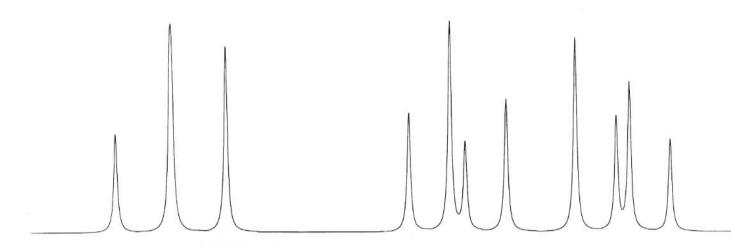


Figure 4. Spectrum produced by close guesses for analysis of ABX case.

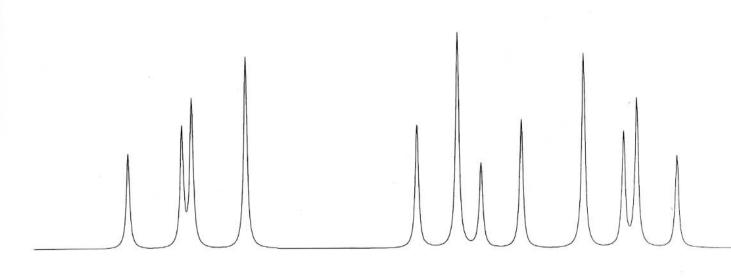


Figure 5. Iterated result of ABX spectrum.

```
ITRCAL
         VIR
                  04/73
NUMBER OF SPINS= 213
SPECTROMETER FREG (MHZ)=
WIDTH OF DISPLAY (HZ)= 90
OFFSET OF DISPLAY (HZ)=
INTENSITY THRESHOLD= 0
CHEMICAL SHIFTS
V(1)= 14.459
V(2)= 34.896
V(3)= 70.67
COUPLING CONSTANTS
J(1,2)= 5.098
J(1,3)= 7
J(2,3) = 7
SET HORIZ DISPLAY SCALE TO 4K.
SET UFRT DISPLAY SCALE TO 1K (READJUST AS NEEDED).
DESIRED VALUE OF RMS ERROR= .01
MAXIMUM # OF ITERATIONS= 10
FNTER VARIABLES
SET 1: V(1)
SET 2: V(2)
SET 3: V(3)
SET
     4: J(1,2)
SET 5: J(1,3)
SET 6: J(2,3)
SET 7:
ARE YOU SURE? Y
ITERATION Ø:
                  0 1 2 3
                        0.681 =8.526
1.077 =13.624
             7.928
  7
     8
  4
            13.110
      6
            14.746
                        0.828 =15.294
  3
      5
            19.929
                       1.411 =20.392
     2
  1
      7
                        0.968 =28.215
0.641 =33.401
  6
            28.730
  4
            33.913
            35.826
                        1.522
  2
                               =36.390
      5
            41.009
                                =41.488
            64.268
                        1.350
                               =63.199
  5
      8
            71.087
                        1.039
                               =70.055
            71.365
                        0.889
                               =71.286
     4
            78.183
                        0.719
                                =78.142
ITERATION 0: 0 1 2
RMS ERROR = 6.07254E-1
ITERATION 1: 0 1 2
RMS ERROR = 2.64898E-2
                     1 2 3
ITERATION 2: 0 1 2
RMS ERROR = 2.54033E-2
ITERATION 3: 0 1 2
RMS ERROR = 2.54033E-2
CHEMICAL SHIFTS
U(1)=
          14.990
           35.017
V(3)=
           69.994
COUPLING CONSTANTS
             5.0128
7.0146
J(1,2)=
J(1,3)=
J(2,3)=
             8.0134
>1
LINE WIDTH= .5
 > T
T
   7
              8.492
       8
                         0.677
                                     8.526
             13.613
                         1.076
                                    13.624
   3
       5
                                    15.294
             20.425
                         1.415
                                    20.392
             28.258
                         0.931
                                    28.215
   4
             33.379
                         0.614
                                    33.401
   2
       5
             36.368
                         1.560
                                    36.390
             41.488
                         0.893
                                    41.488
   5
       8
             63.210
                         1.390
                                    63.199
             70.022
                         1.062
                                    70.055
   S
             71.319
                         0.854
                                    71.286
   1
                         0.691
                                    78.142
 *LOAD TIPRGM
 *G0
```

& PL

B. An A₂B₂ system - ortho-dichlorobenzene

As a second example, let us consider an experimental spectrum of <u>ortho-</u>dichlorobenzene obtained in acetone-d₆ solvent. The chemical shifts and coupling constants for this species will vary somewhat from those reported⁸ for neat ODCB. The spectrum shown in Figure 6 was obtained from a single pulse followed by an accumulation of 8192 points over 40.96 seconds. The region plotted out is the center 50 Hz of this region. Peak printout is as shown, with x's indicating spurious noise peaks. The line which is out of phase is the folded back TMS line.

To analyze this spectrum, the reported shifts were converted to 100 MHz and entered as the guessed values. The spectrum which was produced is shown in Figure 7. Note the degeneracy messages printed during calculation which warn the user to be careful in making line assignments.

To begin iteration, the \underline{I} command is given as shown in the printout on the following pages. An rms error of .01 is entered along with a maximum of ten iterations. The variables sets are then etered. Note that in this case V(1) and V(4) are equivalent as are V(2) and V(3), J(1,2) and J(3,4) and J(1,3) and J(2,4). This statement of equivalence is necessary so that these parameters will not become non-equivalent. The variables J(1,4) and J(2,3) are entered on separate lines to indicate that they are to be varied but not together. Set 7 is terminated with a Return, causing an end to the parameter set entry. The set segments are confirmed by answering Y to ARE YOU SURE?.

The first iteration is merely a calculation based on the guessed parameters. It is accompanied by the warnings regarding degeneracy. Then the line assignments are made based on the peak printout shown in Figure 6. Note the use of the ditto mark to indicate the assignment should be the same as the previously entered one. Iteration begins automatically following the entry of the ditto mark transitions 5-7. Convergence is achieved in three iterations when an rms error of .1010 is obtained. Iteration ceases here because the difference in rms error is less than 3% from one iteration to the next.

The iterated shifts and coupling constants are shown along with the calculated and observed transition frequencies.

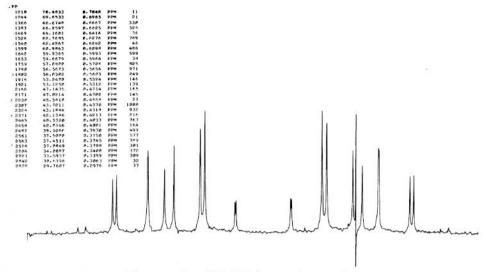


Figure 6. ODCB in acetone-d₆

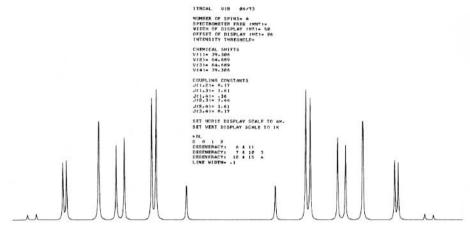


Figure 7. Uniterated ODCB spectrum

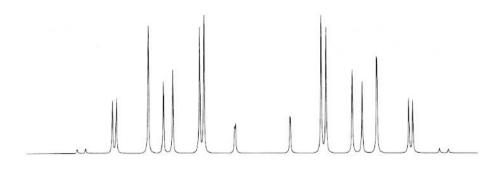


Figure 8. Iterated ODCB spectrum

NUMBER OF SPINS= 4
SPECTROMETER FREQ (MHZ)=
WIDTH OF DISPLAY (HZ)= 50
OFFSET OF DISPLAY (HZ)= 26
INTENSITY THRESHOLD=

CHEMICAL SHIFTS V(1) = 39.306 V(2) = 64.689

V(3)= 64.689 V(4)= 39.306

COUPLING CONSTANTS

J(1,2) = 8.17

J(1,3)= 1.61

J(1,4) = .36

J(2,3) = 7.44J(2,4) = 1.61

J(3,4) = 8.17

SET HORIZ DISPLAY SCALE TO 4K.
SET VERT DISPLAY SCALE TO 1K (READJUST AS NEEDED).

>I I DESIRED VALUE OF RMS ERROR= .01 MAXIMUM # OF ITERATIONS= 10

ENTER VARIABLES

SET 1: V(1) V(4) SET 2: V(2) V(3) SET 3: J(1,2) J(3,4) SET 4: J(1,3) J(2,4) SET 5: J(1,4)

SET 6: J(2,3)

SET 7:

ARE YOU SURE? Y

ITERATION 0: 0 1 2
DEGENERACY: 6 & 11
DEGENERACY: 7 & 10 3
DEGENERACY: 12 & 15 4

29.671 0.060 =29.7607 10 14 29.671 0.064 =" 4 6 30.611 0.146 =30.6936 12 16 33.506 1.280 =33.5937 9 12 33.924 1.325 =34.009 1.485 3 11 37.422 =37.4511 7 13 37.450 0.756 =37.5 10 13 37.450 0.704 39.370 11 14 =39.3066 1.680 4 10 40.263 0.893 =40.0146 4 40.263 0.959 =40.3320 43.286 2.719 =43.1884 1 5 =43.7011 43.754 2.913 8 0.538 47.101 =47.0214 6 13 3 10 47.121 0.243 =47.1425 =" 47.121 0.226 3 7 7 12 56.873 0.226 =53.1250 =" 56.873 0.243 10 12 7 15 56.893 0.278 =53.2470 =" 10 15 56.893 0.259 8 14 2.913 =56.5673 60.240 60.708 2.719 14 16 =57.0800 =59.9363 2 6 63.731 1.853 5 11 64.624 1.680 =60.9862 66.545 1.461 =62.4267 6 15 =" 66.572 1.485 11 12 =66.2597 70.070 1.325 3 70.488 1.280 =66.6748 73.382 0.070 =69.6533 2 10 =" 73.382 0.075 5 10 74.323 0.064 =70.4833 0.068 =" 74.323

```
ITERATION 0:
                         2
DEGENERACY:
                6 & 11
DEGENERACY:
                7 & 10
                         3
DEGENERACY:
               12 & 15
                         4
RMS ERROR =
               1.21419E1
ITERATION
                  0
                     1
DEGENERACY:
                6 & 11
DEGENERACY:
                7 & 10
                         3
DEGENERACY:
               12 & 15
                         4
RMS ERROR =
               1.02095E-1
ITERATION
                  0
DEGENERACY:
                6 & 11
DEGENERACY:
                7
                  &
                    10
                         3
DEGENERACY:
               12 & 15
RMS ERROR =
               1.01065E-1
CHEMICAL SHIFTS
          39.397
V(1)=
V(2)=
          60.821
V(3)=
          60.821
V(4)=
          39.397
COUPLING CONSTANTS
J(1,2) =
            7.9426
J(1,3) =
             1.6150
            0.3581
J(1,4) =
J(2,3) =
            7.5361
J(2,4) =
             1.6150
J(3,4) =
            7.9426
> T
T
  7 14
           29.647
                       0.000
                                 29.760
 10 14
           29.729
                       0.107
                                 29.760
  4
    6
           30.677
                       0.124
                                 30.693
 12 16
           33.594
                       0.592
                                 33.593
  9
    12
           34.061
                       0.616
                                 34.009
  7
    13
           37.530
                       1.416
                                 37.500
  3
    11
           37.564
                       1.455
                                 37.451
    13
 10
           37.612
                                 37.500
                       0.000
 11 14
           39.194
                       1.620
                                 39.306
  4
    10
           40.164
                       0.000
                                 40.014
  4
     7
           40.246
                       1.875
                                 40.332
  1
     5
           43.158
                       2.814
                                 43.188
  5
     8
           43.672
                       3.087
                                 43.701
  3
    10
           47.028
                       0.495
                                 47.142
    13
           47.099
                       0.583
                                 47.021
  6
  3
     7
           47.111
                       0.000
                                 47.142
  7
    15
                                 53.247
           53.113
                       0.292
  7
                                 53.125
    12
           53.113
                       0.290
 10
    15
           53.195
                       0.247
                                 53.247
 10
    12
           53.195
                       0.247
                                 53.125
  8
    14
           56.545
                       3.087
                                 56.567
 14
    16
            57.060
                       2.814
                                 57.080
  2
     6
            59.971
                       1.875
                                 59.936
  5
    11
           61.024
                       1.620
                                 60.986
 11
    12
           62.659
                       0.731
                                 62.426
           62.681
  6
    15
                       0.711
                                 62.426
  3
                                 66.259
     q
           66.162
                       1.233
     3
           66.617
                                 66.674
  1
                       1.185
  2 10
           69.458
                       0.000
                                 69.653
           69.540
  2
     7
                       0.124
                                 69.653
           70.488
                                 70.483
  5 10
                       0.107
           70.570
                       0.000
                                 70.483
  5
     7
>L
LINE WIDTH= .1
```

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- This spectrum was provided by L. F. Johnson of Transform Technology, Inc.

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